8th Grade Flight and Space Lesson: [April 6]

Learning Target:

Air Pressure and its effects on objects around them Cartesian Diver - Boyle's Law Day 1 of 3

Warm Up:

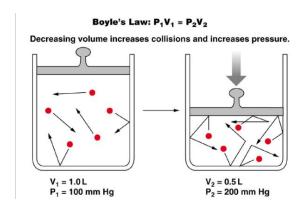
Watch Video: Boyle's Law

Learn about Boyle's law: Khan Academy

Lesson Introduction/Background Information: What is Boyle's Law

- A. It explains why planes fly.
- B. It describes how things grow.
- C. It turns water blue.
- D. It describes how the pressure of a gas tends to increase as the volume of the container decreases.





Gather Materials



Pen Cap
Chewing Gum

(may use clay or other soft pliable material instead if it is available)
Paper Clip
Water bottle filled with water
Water bottle lid





Practice:

Prepare for experiment:

- 1. Find a space to work, like a kitchen table.
- 2. Be sure to have all your materials handy.
- 3. Have towels available to clean up any messes.
- 4. Be sure to have permission before conducting any experiment!

Practice:

Step 1: Make sure you have all of the necessary materials.

Step 2: Fill the plastic bottle with water almost to the top.

Ste 3: Attach a thick layer of chewing gum to the bottom of the plastic pen cap.

Step 4: Use the paper clip to put a single hole in the chewing gum. Make sure it is all the way through.

Step 5: Put the cap in the bottle so that it floats. The clay side should go inside the water bottle first.

Step 6: Put the lid on the bottle and tighten so that it does not leak any air.

Step 7: Squeeze the sides of the bottle.

Step 8: Observe what happens. Repeat step 7 as needed.

Keep your supplies! You will need them for your

day 3 lesson!

Self-Assessment:

Write down your understanding of what Boyle's Law is.

Write down how you think air pressure explains how planes fly.

Additional Practice:

Click on the links below to get additional information and to check your understanding!

Why Boyle's Law is important to NASA

Cartesian Diver experiment video